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$$\frac{23.2 \text{ HC}_2\text{H}_3\text{O}_2 \text{ (aq)}}{23.2 \text{ [HC}_2\text{H}_3\text{O}_2 \text{ (aq)]}} (6.60 \text{ mmol}) = 38.00 \text{ mL}$$
 Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547 Buy Nelson Chemistry 12 Solutions Manual from Nelson's Online Book Store Nelson Chemistry 12 Solutions Manual - Nelson Answers and fully-worked solutions for Practice Questions, Section Questions, Chapter Reviews, and Unit Reviews Sample data and lab reports for in-text investigations, activities, and lab exercises. Nelson Chemistry 12 Lab Answers - plantpono.org Solution books give answers to every question in the textbook and explain the solution in detail. Bellow is a list of popular textbooks and solutions that we sell: *Note: The following is only a small selection of our full collection, if the textbook you are looking for is not on this

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