

# **Explore Learning Stoichiometry Answer Key**

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figuring out the amounts of

reactants and products that take

part in a chemical reaction. To

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begin, check that this equation is shown:  $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$ . Look at the coefficients (such as the "3" in  $3\text{CO}$ ) in front of each substance in the

equation. StoichiometrySE - Name Meaghan Kreh Date Student

... Student Exploration-

Stoichiometry (ANSWER KEY) |

Stoichiometry Practice: Turn off

Show units and Show numerical

result. Click New question, and use

what you've learned to solve

another stoichiometry problem.

Then convert the moles of answer

substance to the desired

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