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constant $0.02/s$ and 27 and $1/s$ at $^{\circ}C$. the a) 428.9 kJ/mol
d) 231 cal/mol For the reaction, the concentration of
that remained after 9 while the initial concentration
was 57.90 M . Calculate the rate a) $-0.0078/\text{min}$ b)
 $0.0726/\text{h}$ c) $1.66/\text{min}$

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